

INTERNATIONAL JOURNAL OF SCIENTIFIC AND UNIVERSITY RESEARCH PUBLICATION

International Journal Of Scientific And University Research Publication

ISSN No 2364/2018

Listed & Index with ISSN Directory, Paris



Multi-Subject Journal



Volum : (3) | Issue : 211 |

Research Paper



ASSESSMENT OF PHYSICO-CHEMICAL PARAMETERS AND WATER QUALITY INDEX (WQI) OF VIDARBHA (IDARBA) RIVER FROM AMRAVATI DISTRICT, MAHARASHTRA, INDIA

G. N. Budhlani

ABSTRACT Vidarbha (Idarba) River passing by Anjansingi and Durgwada towns of Amra Ma¬harashtra is a minor river which flows into the Wardha River.

ماحتى

y Anjansingi and Durgwada towns of Amravati district from Ma¬harashtra is a minor river which flows into the Wardha River. This river is under constant threat of pollution by floral wastes generated on account of pilgrim¬age, agricultural runoff, cattle grazing, faecal contamination and washing of cloths by ladies. The present study was carried out to calculate the Water quality index (WQI) of Vidarbha (Idarba) River to ascertain the water quality status of water for drinking and other purposes. Water quality index was determined on the basis of parameters pH, Total hardness (TH), Total solids (TS), Total alkalinity, Nitrate, Dissolved oxygen, Biological Oxygen demand (BOD), Chemical Oxygen demand (COD) studied for one year during Oct. 2010 to Sept. 2011. The results shows that WQI values of Vidarbha (Idarba) River of Amravati district are above 100 indicating that the source, i.e, Vidarbha River is unsuitable for drinking purposes. The study also revealed that this pollution is due to floral wastes generated on account of pilgrim¬age, agricultural runoff, cattle grazing, faecal contamination and washing of cloths by ladies.

الكلمات الرئيسية:

Vn = estimated value of the nth parameter at a given sampling station.

Sn = standard permissible value of nth parameter

Vio = ideal value of nth parameter in pure water.Ideal value in most

Vio = 0 except in certain parameters like pH and dissolved oxygen. The calculation of quality rating for pH and DO (Vio ≠ 0) is 7.0 and 14.6 mg/l respectively. Unit weight was calculated by a value inversely proportional to the recommended standard values Sn of the corresponding parameters.

Where n = K/SnW

Wn = Unit weight for nth parameter

Sn = Standard value for nth parameters

K = Proportionality constant.

The overall water quality index was calculated by aggregating the quality rating with the unit weight linearly.

$= \Sigma qnWn/\Sigma Wn.Q.I.W$

The suitability of River waters was analyzed on the basis of water quality status (Chatterji and Raziuddin, 2002).

quanty status (Chatter ji and Kaziudum, 2002).					
Unit weight	Recommende	Stand- ards	Parameter	S.No.	
	d agency				
0.2190	ICMR/BIS	6.5-8.5	pН	01	
0.0062	ICMR/BIS	300	Total	02	
			hardness		
			(TH)		
0.0155	ICMR	120	Total	03	
			Alkalinity		
0.0412	ICMR/BIS	45	Nitrate(NO3	04	
0.3723	ICMR/BIS	5.00	Dissolved	05	
			oxygen (DO)		
0.3723	ICMR	5.00	BOD	06	
0.0037	WHO	500	Total solids	07	
			TS		
	•				

able 1:- Drinking Water Standards recommending Agency and T Unit Weights

438 IJSR - INTERNATIONAL JOURNAL OF SCIENTIFIC RESEARCH

Research Paper (All values are expressed in mg/l except pH)

.

Water sources available for drinking and other domestic pur- pose must possess high degree of purity, free from chemical con-2012). Water is also one et al., tamination and micro-organism (Borul of the most important factors for every living organism on this planet. The quality of water is getting vastly deteriorated due to unscientific waste disposal, improper water management and carelessness towards environment, which has also led to scarcity of potable water 2013). Vidarbha (Idarba) et al., affecting the human health (Bhadja river passing by Anjansingi and Durgwada towns of Amravati district from Maharashtra is a minor river which flows into the Wardha River. This river is under constant threat of pollution by floral wastes generated on account of pil- grimage, agricultural runoff, cattle grazing, faecal contamination and washing of cloths by ladies. Water quality index provides a single number that expresses overall water quality at a certain location and time, based on several water quality parameters. The objective of water qual- ity index is to complex water quality data into information that is 2013). The et al., Yogendra understandable and usable by the public (objective of this study is to establish the water quality sta- tus of the river for all seasons of a year. The results reported here will provide base-line data for framing suitable remedial action plan.

:SDHOTMATERIALS AND ME

Water Samples were collected in plastic bottles for physicochemical analysis from 7 sampling stations of Vidarbha (Idarba) River during Oct.2010 to Sept.2011 using standard procedures. Every time pH was monitored at the sampling stations while to- tal hardness, total alkalinity, total solids, nitrates, Dissolved oxy- gen, BOD and COD were analyzed in the laboratory in accord- ance with .,1988 and *et als*tandard methods (Trivedy and Goel,1986; Kodarkar APHA,1995).

I Calculation:OW

Water Quality Index (WQI) indicates the quality of water in terms of index number which represents overall quality of wa- ter for any intended use. It is defined as a rating reflecting the composite influence of different water quality parameters on the overall quality of water quality. For calculation of WQI, selection of parameters has great im- portance. Since selection of too many parameters might widen the water quality index and importance of various parameters -depends on the intended use of water, eight physicochemical pa rameters, namely, pH, total solids, total alkalinity, total hardness, nitrate, Dissolved oxygen (DO), BOD and COD were used to calculate WQI. The calculation of WQI was made using weighted) in et al., 2013 Bhadjaet al., 1972, arithmetic index method (Brown Where)] Vio -Sn) / (Vio -Vn= 100[(qn)the following steps:

qn = quality rating for the nth water quality parameter.

recognized that, like dissolved oxygen, pH also varies in reservoir naturally throughout the day due to the photosynthesis and respiration cy- cles in the presence of algae in water bodies. The pH is meas- ure of the intensity of acidity or alkalinity andthe concentration of hydrogen ion concentration. pH has no direct adverse effects on health; however, higher values of pH hasten the scale forma-tion in water heating apparatus and also reduce germicidal poten- tial of chloride. High pH induces the formation of trihalo meth- ane which is toxic. pH is one of the most important factors that serves as an). From the available et al.,2013index for the pollution (Bhadja data, it appears that the average pH values were 8.57, 8.51 and 8.47 during winter, summer and Monsoon re- spectively. It indicates that pH was slightly alkaline throughout all the seasons and beyond acceptable limit as recommended by BIS (IS,10500;1998). The average values of total hardness of river water were 185.57 mg/L during winter, 184.28 mg/L during summer and 184.85 mg/L during monsoon indicating that total hardness was within the acceptable limit throughout all the sea-Volumeable Issue: 11 | Newember 2015 •Seasonal

ISSN No 2277 - 8179 sons. Hard water causes incrustation in distribution systems and excessive soap consumption (Coleman 1976). The average values of Total solids (TS) of river water samples were 336.71 mg/L,335.85 mg L and 334.42 mg/L during winter, summer and mon- soon respectively. Total solids are measure of suspended and dis-solved solids in a body of water. It appears that river water sam- ples tested were not exceeding the desirable limit et prescribed by BIS (Subin and Aneesha 2011). According to Nayak . (1982) and Ghosh and George (1989) the higher alkalinity al indicates pollution. The value of alkalinity in water provides an idea of natural salts present in water. The cause of alkalinity is the minerals which dissolve in water from soil. The various ionic spe-cies that contribute to alkalinity include bicarbonate, hydroxide, phosphate, borate and organic acids. The average values of total alkalinity were 273.57 mg/L, 274 mg/L and 274 mg/L during winter, summer and monsoon respectively. Total alkalinity was beyond the acceptable limit throughout all the seasons. The aver- age values of Nitrate were 4.94 mg/L, 4.97 mg/L and 6.67 mg/L during winter, summer and monsoon respectively. It indicates that it was within the acceptable limit throughout all the seasons as per the BIS standards. The main source of the formation of ni- trate is the decomposition and biodegradation of organic matters. High nitrates would indicate pollution load. Intrusion of sewage into the natural waters increases levels of nitrate (Manson 1991). High level of DO is normally a sign of healthy river and aver- age values of DO were 7.4 mg/L, 7.3 mg/L and 7.25 mg/L during winter, summer and monsoon respectively. Decrease in DO val- ues can favor anaerobic decomposition of organic wastes (Salle,1974). The average BOD values were 2.85 mg/L, 2.94 mg/L and 2.78 mg/L during winter, summer, and monsoon respectively. The about test 4s meas Water organic load in a body of water. As per BIS, the maximum permissible limit is 5 mg/L and it is within the permissible limit throughout all the seasons as per BIS stand- ards. COD estimates the 2011).COD et al. carbonaceous fraction of organic matter (Arthi values convey the amount of dissolved oxidizable organic matter including the non-biodegradable mat- ters present in it. The average COD values were 11.4 mg/L, 11.4 mg/L and 11.7 mg/L during winter, summer and monsoon respec- tively. Application of WQI is a useful method in assessing water qual- ity at individual sampling station in order to determine its water quality for various beneficial uses. In present study, application of WQI gives us comparative evaluation of water quality at dif- ferent sampling stations during different seasons. The water quality index obtained for the reservoir water system in differ- ent seasons of study period i.e., winter season, summer season and monsoon season are, 113.66, 114.24 and 117.06 respectively which indicate the very poor quality of water (Chatterji and Ra-ziuddin 2002).

able 2: Water Ouality Index (WQI) and status of water qual- T and Raziuddin 2002)

able 2. Water	Quanty muex	(WQI) allu sta
Water quality	Water quality	ity (Chatterji :
status	Index	
	Level	
Excellent water	0-25	
quality		
Good water	26-50	
quality		
Poor water	51-75	
quality		
Very Poor	76-100	
water quality		
Unsuitable for	>100	
drinking		
sonal	variationsT	

of Physicochemical parameters of Vidarbha (Idarba) River water (Oct.2010-Sept.2011)

on r Season ters 8.47 8.51 8.57 pH 01 184.85 184.28 185.57 Total h ardness 02 274 274 273.57 Total A lkalinit y 03 6.67 4.97 4.94 Nitrate(NO3) 04 7.25 7.3 7.4 Dissolv ed oxygen 05 ed oxygen 2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality Index Index Index	Monso	Summe	Winter	Parame	Sr.No
Season Season B.57 pH 01 184.85 184.28 185.57 Total h ardness 02 274 274 273.57 Total A lkalinit y 03 lkalinit y 6.67 4.97 4.94 Nitrate(NO3	on	r		ters	
8.47 8.51 8.57 pH 01 184.85 184.28 185.57 Total h ardness 02 (TH) 274 273.57 Total A lkalinit y 03 6.67 4.97 4.94 Nitrate(NO3) 04 7.25 7.3 7.4 Dissolv ed oxygen 05 (DO) 05 05 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality			Season		
184.85	Season	Season			
ardness (TH) 274 274 273.57 Total A lkalinit y 6.67 4.97 4.94 Nitrate(04 NO3 NO3 NO3 NO3 NO3 NO5 ed oxygen (DO) 2.78 2.94 2.85 BOD 06 NO5	8.47	8.51	8.57	pН	01
(TH) 274 274 273.57 Total A lkalinit y 6.67 4.97 4.94 Nitrate(04 NO3	184.85	184.28	185.57	Total h	02
274 274 273.57 Total A lkalinit y 6.67 4.97 4.94 Nitrate(04 NO3				ardness	
274 274 273.57 Total A lkalinit y 6.67 4.97 4.94 Nitrate(04 NO3					
Ralinit Section Respective Respectiv				(TH)	
Note	274	274	273.57	Total A	03
6.67 4.97 4.94 Nitrate(NO3 7.25 7.3 7.4 Dissolv ed oxygen (DO) 2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				lkalinit	
7.25 7.3 7.4 Dissolv 05 ed oxygen (DO) 2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				у	
7.25 7.3 7.4 Dissolv ed oxygen (DO) 2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality	6.67	4.97	4.94	Nitrate(04
ed oxygen (DO) 2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				NO3	
0xygen (DO) 2.78	7.25	7.3	7.4	Dissolv	05
CDO CDO				ed	
2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				oxygen	
2.78 2.94 2.85 BOD 06 334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality					
334.42 335.85 336.71 Total solids TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				(DO)	
Solids TS	2.78	2.94	2.85	BOD	06
TS 11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality	334.42	335.85	336.71	Total	07
11.7 11.4 11.4 COD 08 117.06 114.24 113.66 Water Quality				solids	
117.06 114.24 113.66 Water Quality				TS	
Quality	11.7	11.4	11.4	COD	08
	117.06	114.24	113.66	Water	
Index				Quality	
Index					
				Index	

T Quality Index of

Vidarbha (Idarba) River water during different Seasons during Oct.2010 to Sept. 2011.

Monsoo	Summer	Winter	Water	1	
n			Quality		
			Index		
			(WQI)		
			Vidarbh		
			a		
117.06	114.24	113.66	(Idarba)		
			River		
			River		
Decults and discussions					

Results and discussion:

The observations on physicochemical parameters of water samples from Vidarbha (Idarba) river passing by Anjansingi and Durgwada of Amravati district studied for one year during Oct.2010-Sept.2011 and Water Quality Index (WQI) of water res- ervoir are presented in Table 1-4. The pH of water body is very important in determination of wa- ter quality since it affects other chemical reactions such as solu- bility and metal toxicity (Agbarie 2009). It should be

استنتاج

From present observations, it may be concluded that WQI values of Vidarbha (Idarba) River passing by Anjansingi and Durg- wada towns of Amravati district of Maharashtra are above 100 indicating that the source, i.e, Vidarbha River water is unsuitable for drinking purposes. This pollution is due to floral wastes generated on account of pilgrimage, agricultural runoff, cattle grazing, faecal contamination and washing of cloths by ladies. Therefore, people in these areas have high potential risk of contracting water borne diseases if they use this river water for drinking purposes. Therefore, it is recommended that river water should not be used for domestic purposes without treatment.

ref str

Agbaire, P.O. and Obi, C.G. (2009): Seasonal Variations of Some Physicochemical Properties of River Ethiope Water in Abraka, Nigeria. J. Appl. Sci. Enviorn. Manage 13(1) 55-57. • Alam Mohammad and Pathak, J.K.(2010): Rapid Assessment of Water Quality Index of Ramganga River, WesternUtter Pradesh (India) Using a Computer Programme, Nature and Science, 8(11), • APHA Manual (1995): Standard methods for the examination of water and waste water, APHA, AWWA, WPET, Washington DC U.S.A., 19th edition. · Bhadja Poonam and Vaghela Ashokkumar.2013. Assessment of physico-chemical parameters and water quality index of Res- ervoir water. International Journal of Plant, Animal and Environmental Sciences,3(3):89-95 • Chaterjee,C. and Raziuddin, M.2002. Determination of water quality index (WQI) of a degraded river in Asanol Industrial area, Raniganj, Burdwan, West Bengal. Nature, Environment and pollution Technology,1(2):181-189. • Chatterjee, P.R., Chinmoy Chatterjee and M. Raziuddin. 2007. Impact of human activity on water quality of a lentic water body in Asansol. Nature. Environ. Poll. Technol., 6(1):59-62. • ICMR. 1975. Manual of standards of Quality for Drinking Water Supplies. Indian Council of Medical Research, Govt. of India, New Delhi, • Jain, C. K., Bhatia, K. K. S. and Vijay, T. (1997): Ground water quality in coastal region of Andhra Pradesh, Indian J. Env. HLTH, Vol. 39(3), PP. 182-192. • Kodarkar, M.S. Diwan, A.D., Murugupan, N., Kulkarni, K.M. and -Anuradha Ramesh. 1998. Methodology for water analysis ((Physico chemical, Biological and Microbiological). Indian, Assoc. Aquatic biologists, Hyderabad Publication, No. 2. • Manja, S.K. and Kaul, R.K. (1992): Efficiency of simple test for Bacteriological quality of water, Journal IAEM, 19: (18-20). • Manja, S.K. and Kaul, R.K. (1992): Efficiency of simple test for Bacteriological quality of water, Journal IAEM, 19:(18-20). • Mishra, Arunabh and Bhatt, Vasishta.(2008): E.J.Chem.,5(3) 487-492. • Murali,,M and T.Indira.2004. Some studies on ground water contamination due to integrated low cast sanitation in slum areas of Vishakapatnam. Nature Environ. Poll. Techno.3(3)249-253. • Murali,, M and T.Indira.2004. Some studies on ground water contamination due to integrated low cast sanitation in slum areas of Vishakapatnam. Nature Environ. Poll. Techno.3(3)249-253. • Trivedy, R. K. and Goel, P. K. (1986): Chemical and Biological methods for water pollu- tion studies, Environmental publications, Karad. Yogendra K. and Puttaiah E.T.(2008). Determination of Water Quality Index and Suitability of an Urban Waterbody in Shimoga Town, Karnataka. • WHO (1993): Guideline for Drinking water qualiy, 2nd edition, Vol 1,PP 188.



IJSURP Publishing Academy International Journal Of Scientific And University Research Publication Multi-Subject Journal

Editor.

International Journal Of Scientific And University Research Publication



www.ijsurp.com